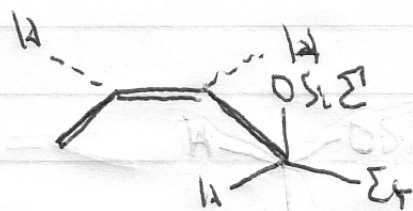
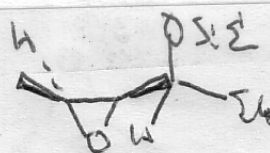


Chem 131 (231) A; Solns to Prob Set 11

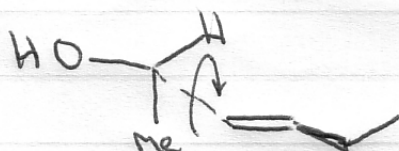
(a)



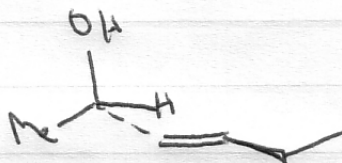
MCPBA



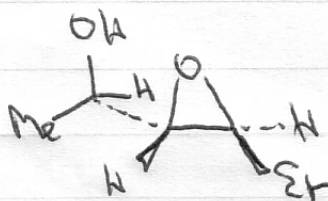
(b)



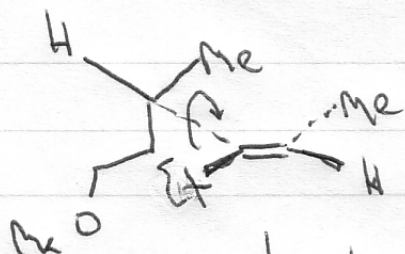
rot'n



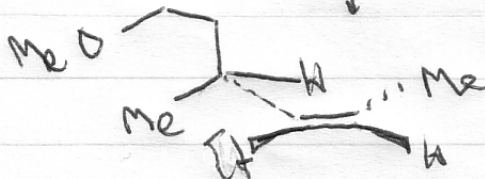
↓ MCPBA (Went best approx.)



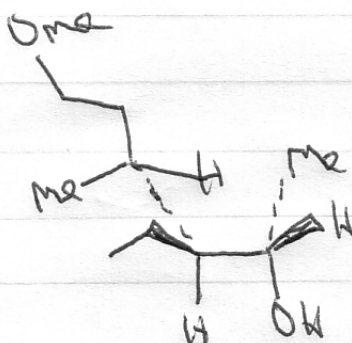
(c)



rot'n

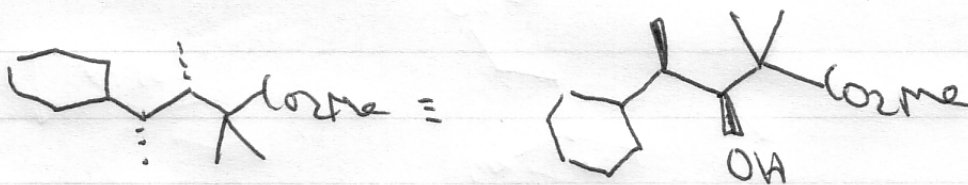
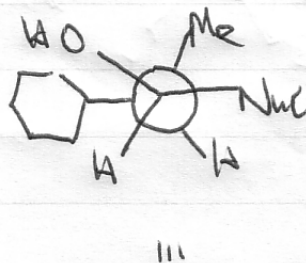
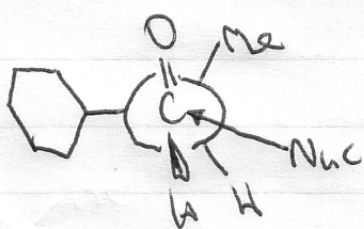
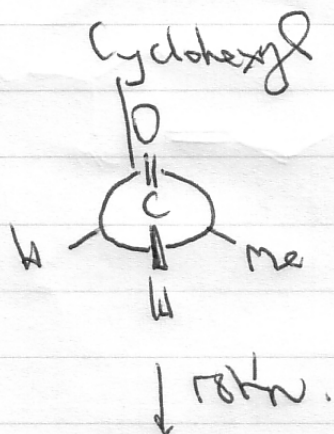
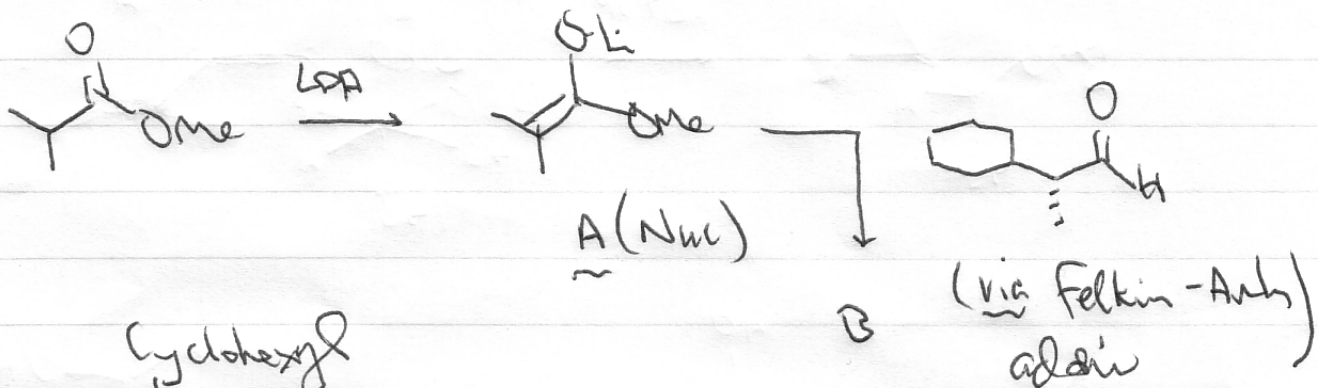


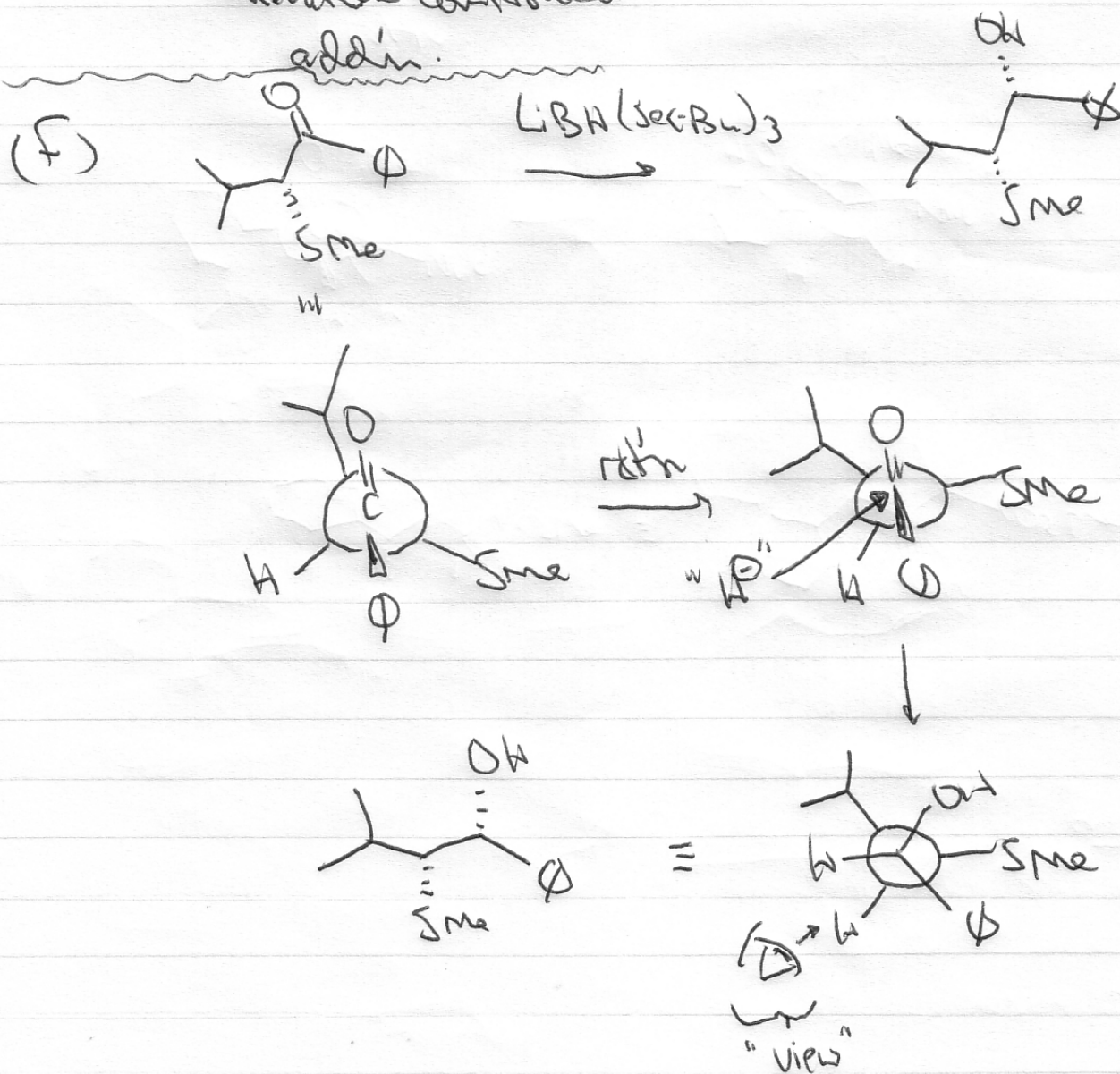
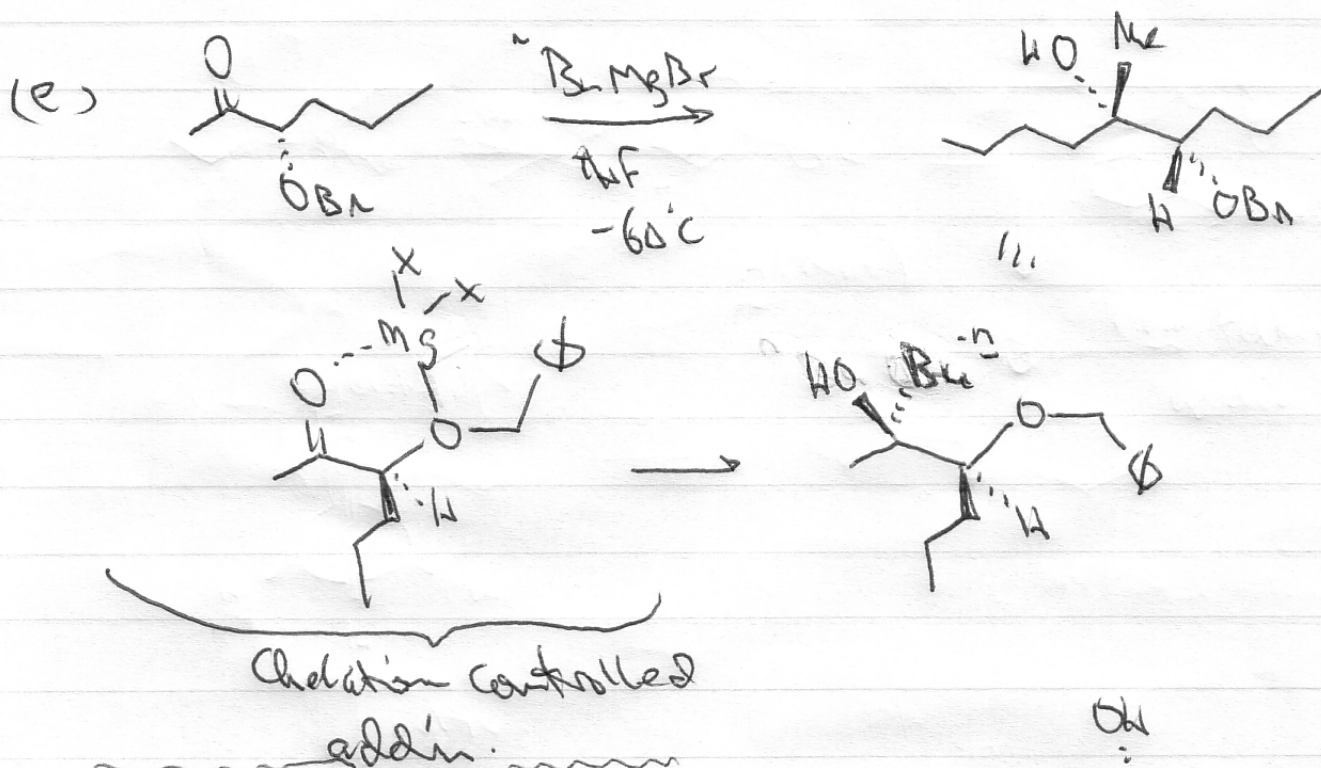
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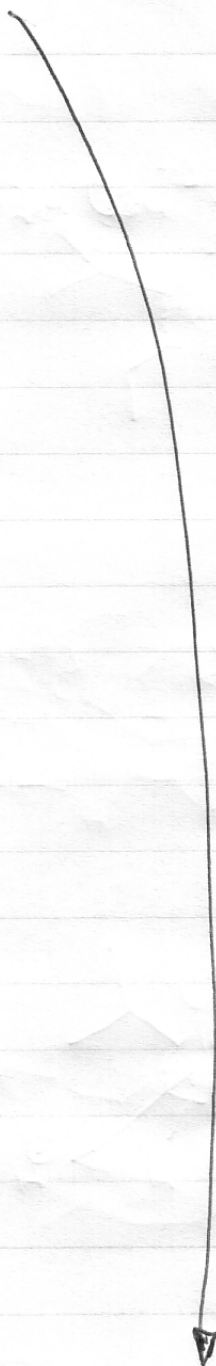
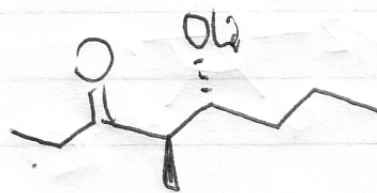
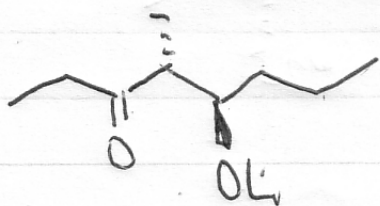
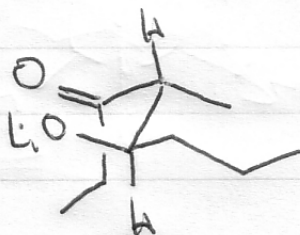
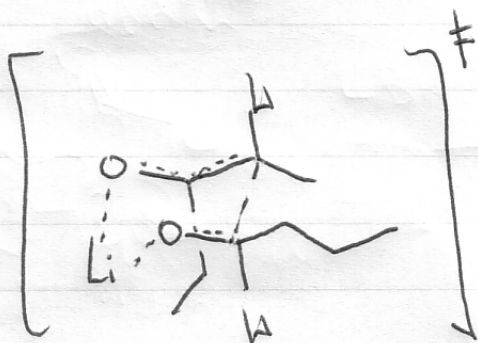
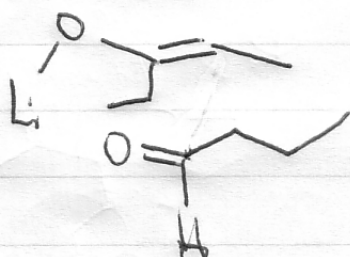
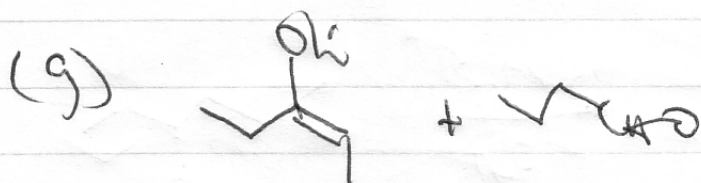


★ make a model & draw so that the longest chain is in the plane of the page. Do so w/ the other cases too.

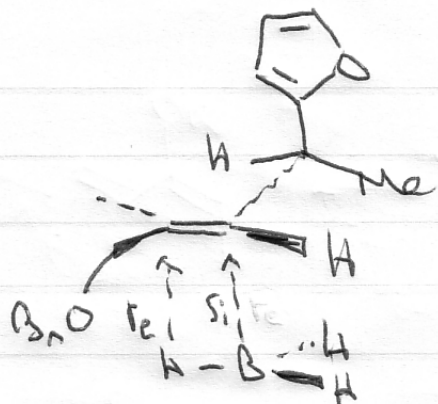
(d)







2.



3. $70/30 \Rightarrow \underline{40\%} [70-30] \text{ de}$

4. $\% \text{ de} = 89$

$$P_1 + P_2 = 100$$

$$P_2 - P_1 = 89$$

$$\underline{2P_2 = 189 \Rightarrow P_2 = 94.5\%}$$

$$\therefore \underline{P_1 = 5.5\%}$$

5. $\left. \begin{array}{l} P_1/P_2 = 10 \\ P_1 + P_2 = 100 \end{array} \right\} \Rightarrow$

$$P_2 = \frac{100}{11} \approx \underline{9.1\%} = P_2$$

$$\therefore \underline{P_1 = 90.9\%}$$

$$\begin{aligned} \Delta \Delta G^\ddagger &= RT \ln 10 = RT \cdot 2.303 (\log 10) \\ &= 2.303 RT = (2.303)(1.987)(298) \\ &\approx 1364 \text{ cal mol}^{-1} \end{aligned}$$

$$\underline{\Delta \Delta G^\ddagger \approx 1.4 \text{ Kcal mol}^{-1}}$$